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# Pseudochirality<sup>1,2</sup>

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Abstract: An algebraic description of pseudochirality (pseudoasymmetry) is presented which is analogous to that given of regular chirality by Wheland. It is found that a chemical structure is pseudochiral if it lacks certain kinds of symmetry based on combinations of chirality reversal operations and point group operations. Pseudochirality groups are defined which include these operations. The derivation of these groups for chemical structures with one or more sets of chiral ligands is described, and a tabulation of general classes of pseudochirality groups based on the usual classes of point groups is given.

The concept of pseudoasymmetry is reasonably well known as a special case of stereoisomerism.<sup>4</sup> It is well known that carbon atoms with four different substituents are chiral, and that the enantiomeric forms can be interconverted by reflection in any mirror plane. However, if two of the substituents are themselves enantiomeric chiral ligands, two isomeric forms result which are meso and cannot be interconverted by reflection in a mirror plane. Such a pair of stereoisomers have been termed a pseudoasymmetric pair.<sup>5</sup> A symbolic example of such a pair is 1a and 1b, a real ex-



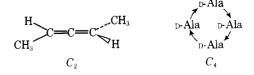
ample is meso-pentaric acid. These are diastereomers and differ in physical properties such as melting point. In all chemical structures in this paper, chiral ligands are designated by open circles or by letters which are themselves chiral in two dimensions. The enantiomeric ligands are then designated by filled circles or reversed letters. This notation has been used extensively by Prelog.<sup>6-9</sup>

An important distinction must be made between pseudoasymmetric structures (1a and 1b) typically termed "meso" and structures such as 2 which exist in only one



achiral meso form. Such a structure will be designated meso from now on, while the pseudoasymmetric structures will be termed pseudoasymmetric or pseudochiral. This is a departure from standard terminology which designates all such structures as meso. The intrinsic difference between the pseudochiral and meso situations will be established later in this paper.

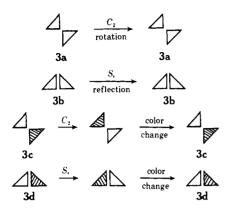
The purpose here is to give an algebraic description of this phenomenon analogous to that for regular chirality. A rigid structure is chiral if it lacks an alternating axis of symmetry.<sup>10</sup> It may, however, have pure rotation symmetry, such as 1,2-dimethylallene ( $C_2$  symmetry) or cyclotetra-Dalanine (in a  $C_4$  conformation).



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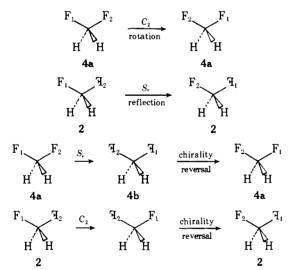
#### Symmetry Operations

To get at the desired algebraic description, an apparent digression must be made. Consider the configurations of triangles 3a-d and the effect of various symmetry operations on them. Configuration 3a is invariant to a  $C_2$  rotation as shown. Configuration 3b is invariant to reflection in a verti-



cal plane. These are the two types of symmetry operations traditionally encountered. Two more will now be defined. Configuration **3c** is not invariant to a  $C_2$  rotation since the triangles are colored differently. However, this configuration is invariant to the sequence of operations shown. First, a  $C_2$  rotation is done followed by a color change operation which colors the white triangle black and the black triangle white. This combined operation will be designated  $\underline{C}_2$ . Similarly, the sequence of reflection and color change leaves configuration **3d** unchanged. This operation will be designated  $\underline{S}_v$ . Thus a total of *four* symmetry operations are defined. This situation (of four types of symmetry operations) has been well studied and is termed *dichromatic antisymmetry*.<sup>11</sup>

Now a similar sequence will be done on the structures 2, 4a, and 4b. Structure 4a is invariant to a  $C_2$  rotation, and



structure 2 is invariant to a  $S_{\nu}$  reflection (aside from ligand numbering). However, structure 4a is also invariant to the sequence of operations shown. First, 4a is reflected in the plane  $S_{\nu}$  which reverses the configuration of the chiral ligands. The operator, *chirality reversal*, is performed which reverses the chirality of all ligands. This changes the ligands back to their original configuration. This sequence of operations will be symbolized  $S_{\nu p}$ . Similarly, the sequence of rotation and chirality reversal leaves 2 unchanged. This combination will be symbolized  $C_{2p}$ . These two combination operations will be termed *pseudochirality operations*.

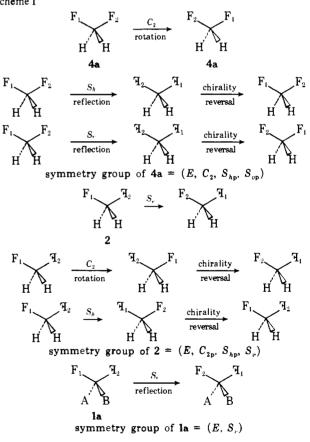
Some further clarification of the effect of these pseudochirality operations may be helpful. A  $C_{np}$  operation in its simplest form reverses the chirality of all chiral ligands in a structure, while it is simply a  $C_n$  rotation for the rest of the structure. Therefore a structure with a  $C_{np}$  element of symmetry is invariant to the  $C_n$  rotation except that all (+)chiral ligands are taken to (-)-chiral ligands of the same constitution. A  $C_{np}$  axis (for a set of chiral ligands) cannot pass through any of the ligands in the set, and any structure with a  $C_{np}$  element of symmetry has an equal number of (+)- and (-)-chiral ligands. An  $S_{np}$  rotation-reflection operation leaves all (+)-chiral ligands as (+)-chiral ligands, while it is simply an  $S_n$  rotation-reflection for the rest of the structure. A structure with an  $S_{np}$  element of symmetry is invariant to the  $S_n$  operation except that all (+)-chiral ligands are taken into (+)-chiral ligands of the same constitution. An  $S_{np}$  axis can pass through chiral ligands, and there are no restrictions on the relative number of (+)- and (-)-ligands.

Comparison of the color change and chirality change operations shows one significant difference. Reflection does not change color, while it does change chirality. This means that the reflection operation in the color change case corresponds to the reflection plus chirality change operations in the chirality change case.

## **Group Theoretic Properties**

It is well known that all the symmetry operations of a structure form a group, hence the group properties of these pseudochirality operators will be investigated.

In Scheme I are shown three structures and the symme-Scheme I



try operations to which they are invariant. It is a simple matter to verify that these operations form the groups indicated (hereafter termed pseudochirality groups). For structures **4a** and **2**, the group is isomorphic to the symmetry point group  $C_{2\nu}$ . In fact, these structures would have  $C_{2\nu}$ 

symmetry if the ligands were not chiral. Structure 1a would have  $C_s$  symmetry without chiral ligands. The problem now is to determine how these groups can be derived and what relation they have to the problem at hand.

The mathematical problem here can be concisely stated. The group that includes *all* of the symmetry and pseudochirality operators is the direct product  $C \times S$ . S is the symmetry point group, and C is the chirality change group. In the present example, C is isomorphic to the cyclic group of order two  $C_2$ . In the general case, C will be the direct product of cyclic groups of order 2. The direct product  $C \times S$ has the property shown diagrammatically in eq 1. The ar-

$${}^{1_{c}} \bigcirc C \stackrel{i_{c}}{\rightleftharpoons} C \times S \stackrel{p_{s}}{\longleftrightarrow} S \bigcirc {}^{1_{a}}$$
(1)

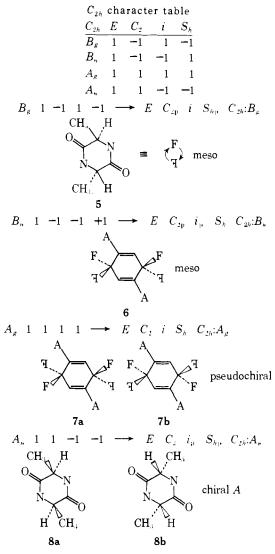
rows refer to homomorphisms between the groups. The homomorphism  $i_s$  is not unique. Composing the homomorphisms shown in eq 1 gives the following results. Injecting Cinto  $C \times S$  (*i*<sub>c</sub>) followed by the projection  $p_c$  gives the same result as the identity mapping of C onto itself,  $1_c$ . Similarly,  $i_s$  followed by  $p_s$  gives  $1_s$ . The significance of all this to the problem at hand can now be stated. The possible pseudochirality groups correspond to the possible injections  $i_s$ . The desired group is the image in  $C \times S$  of  $i_s$ . The image of  $i_s$  in  $C \times S$  cannot contain the operation which simply reverses the chiral ligands without rotation or reflection. Such an operation clearly cannot be a symmetry operation for a structure. Stated differently, C and S are disjoint except for the identity. This is a property of direct products. Now the composition of  $i_s$  with  $p_c$  gives a projection of S onto C. It has already been noted that C is isomorphic to  $C_2$ , hence each pseudochirality group determines a homomorphism of S onto  $C_2$ . This property suggests an easy way of deriving these groups, using chemically familiar group theoretic methods. Each homomorphism of a group onto  $C_2$  determines a one-dimensional irreducible representation of the group. Thus the desired pseudochirality groups correspond to the one-dimensional, real irreducible representations of the symmetry point group.

The derivation of the pseudochirality groups obtainable from  $C_{2h}$  is sketched in Scheme II along with examples. The desired groups are obtained by "imposing" the onedimensional irreducible representation on the point group. For example, consider the  $B_g$  representation. Imposition maps E and i to +1 and  $C_2$  and  $S_h$  to -1. This corresponds to the pseudochirality group (E,  $C_{2p}$ , i,  $S_{hp}$ ). This group is given the unique designation  $C_{2h}$ :  $B_g$ . The structure with this symmetry is meso 5 (one isomeric form). Similarly the  $B_{\mu}$  representation yields a pseudochirality group and a meso structure 6. Now imposition of the  $A_g$  representation maps all the group elements to +1. However, it must be remembered that chiral ligands are present. This group includes no pseudochirality operators and describes the symmetry of a pseudochiral structure with  $C_{2h}$  point group symmetry. The two pseudoenantiomeric forms (7a, 7b) of such a structure are shown. The reason for using the term pseudochiral rather than pseudoasymmetric is now apparent. This example satisfies the requirement for being "pseudoasymmetric" yet has nontrivial rotation symmetry, hence the term pseudochirality is preferred. This parallels the reason for using the term chirality rather than asymmetry to describe traditional enantiomerism.<sup>10,12</sup>

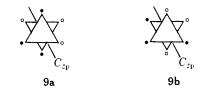
The final representation is  $A_{u}$ , and imposition yields the group  $(E, C_2, i_p, S_{hp})$ . This group includes no alternating axis of symmetry, hence any structure with this symmetry must be chiral and exist in two enantiomeric forms, as does the example (**8a**, **8b**).

At this point, the possible types of isomerism can be de-

Scheme II



scribed. The type of isomerism exhibited by a chemical structure is determined by the kinds of symmetry the structure has. This information is contained in Table I along with examples. Structures 9 and 10 can be thought of as oc-



tahedrally coordinated metals with two tridentate ligands, themselves containing chiral ligands. In all cases, the isomeric forms are interconverted by operations which are not included among the symmetry operations. This gets complicated in the latter case (chiral C), and the interconversions of the four isomers (**10a-d**) are indicated.

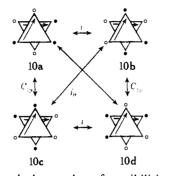
The desired definition of pseudochirality can now be given. A structure is pseudochiral if it lacks  $C_{np}$  and  $S_{np}$ axes of symmetry (for a single set of chiral ligands, vide infra). This definition resembles Wheland's familiar definition of chirality (lack of an  $S_n$  axis). A structure which satisfies both these criteria will be both chiral and pseudochiral and exist in four isomeric forms. An example is provided by structures **10a-d** (Scheme III).

A generalization is possible by allowing more than one set of chiral ligands. Such sets can be constitutionally different or be situated in sites that are never interchanged. As Table I. Types of Isomerism

Name	Cn	Cnp	Sn	Snp	No. of isomers <sup>a</sup>	Kind of isomers	Examples
Meso	Yes	Yes	Yes	Yes	1	Meso	2, 5, 6
Pseudochiral	Yes	No	Yes	No	2	Pseudoenantiomers	1, 7
Chiral A	Yes	No	No	Yes	2	Enantiomers	4.8
Chiral B	Yes	Yes	No	No	2	Enantiomers	9
Chiral C	Yes	No	No	No	4	Pseudoenantiomers and enantiomers	10

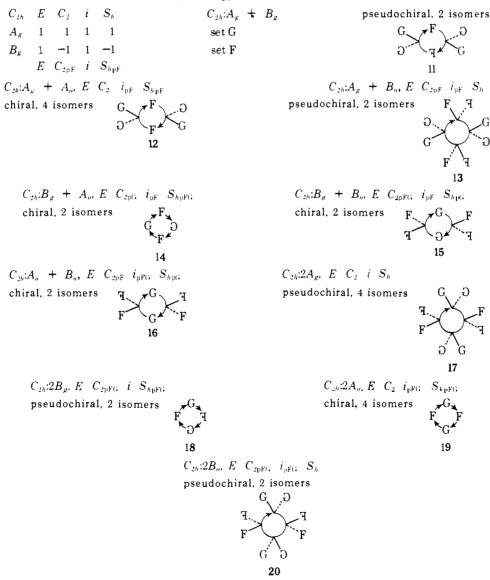
<sup>a</sup> This is the number of isomers which have the same pseudochirality symmetry group.

Scheme III



might be expected, the number of possibilities increases significantly. However, the method of determining the pseudochirality groups and types of isomerism is easily derived

Scheme IV. Derivation of Double Pseudochirality Groups from  $C_{2h}$ 



from the simple case. In the case of n sets of chiral ligands, the group C in (1) is the direct product of n cyclic groups of order two. The desired pseudochirality groups are therefore in correspondence with all the possible sums of n one-dimensional irreducible representations of the symmetry

point group. The possible double pseudochirality groups for  $C_{2h}$  are derived in Scheme IV in an abbreviated form, except for the first one. The two representations  $A_g$  (for the set of F ligands) and  $B_g$  (for the G set) are imposed on  $C_{2h}$ . The resulting group,  $C_{2h}:A_g + B_g$ , is shown and an example given. The operator  $C_{2pF}$  is a  $C_2$  rotation followed by a chirality reversal operation on the F ligands only. Only one iso-

Using these methods, the possible single and double pseu-

dochirality groups have been derived and are listed in Table II. Only one representative group from each class of sym-

meric form of each structure is shown.

Table II. Possible Pseudochirality Groups<sup>21</sup>

Super		Туре	Isomers	Super- group	Derived group name	Туре	Isomers
$\overline{C_1}$	(1) $C_1:A$	с	4a		$C_{2h}:A_u + B_u$	c	2 <i>n</i>
	(2) $C_1 : 2A$	с	8		$C_{2h}:A_g + B_g$	p	20
$C_s$	(1) Single				$C_{2h}: A_g + A_u$	c	4 <i>p</i> 29
- 3	$C_{s}:A'$	p	2 <i>b</i>		$C_{2h}: A_g + B_u$ $C_{2h}: B_g + A_u$	p	$\frac{2^{4}}{2^{r}}$
	$C_s:A''$	c	2		$C_{2h}:B_g + B_u$	c c	$\frac{2^{\prime}}{2^{s}}$
	(2) Double				-	C	25
	$C_s: 2A'$	р	4	C <sub>3h</sub>	(1) Single		
	$C_s: 2A''$	с	4 4		$C_{3h}:A'$	p	2
	$C_{s}:A'+A''$	с	4		$C_{3h}^{''}(A'')$ (2) Double	с	2
C <sub>i</sub>	(1) Single				$C_{3h}:2A'$	p	4
	$C_i:A_g$	p	2		$C_{ab}:2A''$	c	4
	$C_{i}:A_{u}$	с	2		$C_{3h}^{''}:A' + A''$	с	4
	(2) Double $C_i: 2A_g$	n	4	D₃h	(1) Single		
	$C_i: 2A_u$	p c	4	- 371	$D_{3h}:A_1$	p	2
	$C_i:A_g + A_u$	c	4		$D_{3h}^{3n}:A_{2}^{\prime}$	p	2
~	-				$D_{3h}:A_{1}$	с	2
22	(1) Single	с	4		$D_{3h}:A_2$	m	1
	$\begin{array}{c} C_2:A\\ C_2:B \end{array}$	c	$\frac{1}{2^c}$		(2) Double		
	(2) Double	C	-		$\begin{array}{c} D_{3h}:2A_{1}'\\ D_{3h}:2A_{2}'\\ \end{array}$	p	4 2
	$C_{2}:2A$	с	8		$D_{3h} : 2A_{1}$ $D_{3h} : 2A_{1}$	p c	4
	$C_{2}:2B$	с	4		$D_{3h}: 2A_2$	p	2
	$C_2:A+B$	с	4		$D_{3h}^{3n}:A_{1}^{'}+A_{2}^{'}$	p	2 2 4
D₄	(1) Single				$D_{3h}:A_{1}'+A_{2}'$ $D_{3h}:A_{1}'+A_{1}''$	c	
- 4	$D_4:A_1$	с	4		$D_{3h}:A_{1}'+A_{2}''$	р	2
	$D_4:A_2$	с	2		$D_{3h}:A_{2}' + A_{1}''$	с	2
	$D_4:B_1$	с	2		$D_{3h}^{J''}:A_{2}^{J''}+A_{2}^{J''}$ $D_{3h}^{J''}:A_{1}^{J''}+A_{2}^{J''}$	с	2
	$D_4:B_2$	с	2		$D_{3h}:A_1 + A_2$	с	2
	(2) Double $D_4:2A_1$	с	8	$D_{4h}$	(1) Single		
	$D_4:2A_1$ $D_4:2A_2$	c	4		$D_{4h}:A_{1g}$	р	2
	$D_4:2B_1$	c	4		$D_{4h}:A_{2g}$	m	1
	$D_{4}^{-}:2B_{2}^{+}$	с	4		$D_{4h}:B_{1g}$	m	1
	$D_4:A_1 + A_2$	с	4		$D_{4h}:B_{2g}$	m	1
	$D_4:A_1+B_1$	с	4		$D_{4h}:A_{1u}$	c m	2 1
	$D_4:A_1+B_2$	c c	4 4		$D_{4h}:A_{2u}$ $D_{4h}:B_{1u}$	m	1
	$D_4:A_2 + B_1$ $D_4:A_2 + B_2$	c	2		$D_{4h}:B_{2u}$	m	1
	$D_4:B_1^2 + B_2^2$ $D_4:B_1^2 + B_2^2$	c	2		(2) Double		
-					$D_{4h}: 2A_{1g}$	р	4
$C_{2\nu}$	(1) Single	-	2		$D_{4h}:2A_{2g}$	p	2
	$C_{2\nu}:A_1$	p c	2		$D_{4h}: 2B_{1g}$ $D_{4h}: 2B_{2g}$	p p	2
	$C_{2\nu}:A_2 C_{2\nu}:B_1$	m	$1^d$		$D_{4h}: 2D_{2g}$ $D_{4h}: 2A_{1u}$	р с	2 2 2 4
	$C_{2\nu} = C_{2\nu}$	m	$1^{e}$		$D_{4h}$ : 2A <sub>2U</sub>	p	2
	(2) Double				$D_{4h}: 2B_{1u}$	p	2
	$C_{2\nu}: 2A_{1}$	р	4		$D_{4n}: 2B_{2u}$	p	2
	$C_{2\nu}$ : 2 $A_2$	с	4 2		$D_{4h}A_{1g} + A_{2g}$	p	2 2
	$C_{2\nu}: 2B_1$ $C_{2\nu}: 2B_2$	p	2		$D_{4h}:A_{1g} + B_{1g}$ $D_{4h}:A_{1g} + B_{2g}$	p p	2
	$C_{2\nu}: 2B_2$ $C_{2\nu}: A_1 + A_2$	p c	4		$D_{4h}:A_{1g} + A_{1u}$	р С	4
	$C_{2\nu}:A_1 + B_1$	p	2		$D_{Ah}:A_{Aa} + A_{Aa}$	р	2
	$C_{2\nu}:A_1 + B_2$	p	2		$D_{Ah}:A_{1g} + B_{1u}$	p	2
	$C_{2V}:A_{2}+B_{1}$	с	2		$D_{Ah}:A_{1a} + B_{2h}$	p	2
	$C_{2\nu}: A_{2} + B_{2}$	с	2 2 <sup>22</sup>		$D_{4h}:A_{2g} + B_{1g}$	m	1 1
	$C_{2\nu}:B_{1}+B_{2}$	с	2		$D_{4h}:A_{2g} + B_{2g}$ $D_{4h}:A_{2g} + A_{1u}$	m c	2
- 3V	(1) Single				$D_{4h}:A_{2g} + A_{2u}$	c	2
50	$C_{3V}:A_1$	р	2		$D_{4h}:A_{2g}+B_{1u}$	m	1
	$C_{3\nu}:A_2$	с	2		$D_{Ah}:A_{2\sigma}+B_{2\mu}$	m	1
	(2) Double	-	4		$D_{4h}:B_{1g}+B_{2g}$	m	1
	$C_{3\nu}: 2A_1$ $C_{3\nu}: 2A_2$	p c	4		$D_{4h}:B_{1g}+A_{1u}$	c	2 1
	$C_{3\nu}: A_1 + A_2$	c	4		$D_{4h}:B_{1g} + A_{2u}$	m c	2
-		-			$D_{4h}^{h}:B_{1g}^{g} + B_{1u}^{u}$ $D_{4h}:B_{2g}^{g} + A_{1u}^{u}$	c	2
2h	(1) Single	-	$2^{f}$		$D_{Ah}:B_{1g}+B_{2u}$	m	1
	$C_{2h}:A_g$	p	25 18		$D_{ih}:B_{ag} + A_{au}$	m	1
	$C_{2h}:B_g$	m	$2^{h}$		$D_{Ah}:B_{2\sigma}+B_{1\mu}$	m	1
	$C_{2h}: A_u$	c m	$\frac{2^{n}}{1^{i}}$		$D_{4h}:B_{2g} + B_{2u}$	c	2 2
	$\begin{array}{c} C_{2h}:B_{u} \\ (2) \text{ Double} \end{array}$	111	1		$D_{4h}:A_{1u} + A_{2u}$ $D_{4h}:A_{1u} + B_{1u}$	c c	2
	$C_{2h}:2A_g$	р	4 <i>i</i>		$D_{4h}:A_{1u} + B_{1u}$ $D_{4h}:A_{1u} + B_{2u}$	c	2
	$C_{2h}$ : $2B_g$	p	$2^k$		$D_{Ah}:A_{2\mu}+B_{1\mu}$	m	1
	$C_{2h}: 2A_{\mu}$	с	41		$D_{4h}:A_{2u} + B_{2u}$	m	1
	$C_{2h}: 2B_u$	р	2 <i>m</i>		$D_{4h}:B_{1u} + B_{2u}$	m	1

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uper- roup	Derived group name	Type	Isomers	Super- group	Derived group name	Type	Isomers
$D_{2d}$	(1) Single			I <sub>h</sub>	(1) Single		
	$D_{2d}:A_1$	p	2		$I_h:A_g$	р	2
	$D_{2d}:A_2$	m	1		$I_h:A_u$	с	2
	$D_{2d}:B_1$	с	2		(2) Double		
	$D_{2d}:B_2$	m	$1^{u}$		$I_h: 2A_g$	p	4
	(2) Double	-	4		$I_h: 2A_u$	c	4
	$D_{2d}:2A_1$	p	4		$I_h:A_g + A_u$	с	4
	$D_{2d}$ : 2A <sub>2</sub>	p	2 4	Т	(1) Single		
	$D_{2d}: 2B_1$ $D_{2d}: 2B_2$	c	2		T:A	с	4
	$D_{2d} : 2D_2$ $D_{2d} : A_1 + A_2$	p	2		(2) Double		
	$D_{2d}:A_1 + B_1$	p c	4		T:2A	с	8
	$D_{2d}:A_1 + B_2$	p	2	0	(1) Single		
	$D_{2d}:A_{2} + B_{1}$	C P	2	0	(1) Single $O:A_1$	6	4
	$D_{2d}:A_2 + B_2$	c	2		$O:A_1$ $O:A_2$	c c	2
	$D_{2d}:B_1 + B_2$	с	2		(2) Double	t	2
					$O:2A_1$	с	8
3d	(1) Single		2		$O:2A_2$	c	4
	$D_{3d}:A_{1g}$	p	2		$O:A_1^2$ $O:A_1 + A_2$	c	4
	$D_{3d}:A_{2g}$	m	$\frac{1}{2}$		0.211 . 212	C	-
	$D_{3d}:A_{1u}$	с	2 1	Ι	(1) Single		
	$\begin{array}{c} D_{3d}:A_{2u}\\ (2) \text{ Double} \end{array}$	m	1		I:A	с	4
		n	4		(2) Double		
	$D_{3d}: 2A_{1g}$ $D_{3d}: 2A_{2g}$	p	2		I:2A	с	8
	$D_{3d}$ : $2A_{2g}$ $D_{3d}$ : $2A_{1u}$	p c	2 4	C	(1) 641-		
	$D_{3d}$ : $2A_{1u}$ $D_{3d}$ : $2A_{2u}$		2	C∞v	(1) Single		
	$D_{3d}$ , $2A_{2u}$	p p	$\frac{2}{2}$		$C_{\infty_{\nu}}: \Sigma^+ (\text{ref } 23)$		-
	$D_{3d}:A_{1g} + A_{2g}$ $D_{3d}:A_{1g} + A_{1u}$	р С	4		$C_{\infty v}: \Sigma^{-}$	с	2
	$D_{3d}:A_{1g} + A_{2u}$	p	2		(2) Double		
	$D_{3d}:A_{2g} + A_{1u}$	р С	2		$C_{\infty_{V}}:2\Sigma^{-}$	С	4
	$D_{3d}:A_{2g} + A_{2u}$	c	2		The other combinations are	not possible	
	$D_{3d}:A_{1u} + A_{2u}$	c	2	Dark	(1) Single		
		C C	-	- ~n	$D_{\infty h}: \Sigma_g^+$ Not possible		
	(1) Single				$D_{\infty h}: \Sigma_g^{-}$	m	1 <i>w</i>
	$S_4:A$	р	2 <sup>t</sup>		$D_{\infty h}: \Sigma_{u}^{+}$ Not possible	111	1
	$S_4:B$	с	2				22
	(2) Double				$D_{\infty h}: \Sigma_{\mu}^{-}$	с	$2^{x}$
	$S_4:2A$	р	4		(2) Double		2
	S <sub>4</sub> :2B	с	4		$D_{\infty h}: 2\Sigma g^{-}$	р	2
	$S_4:A + B$	с	4		$D_{\infty h}: 2\Sigma_u^-$	с	4
	(1) Single				$D_{\infty h}: \Sigma_g + \Sigma_u^-$	с	2
d	$T_d:A_1$		2		The other combinations are a	not possible	
		p c	2	$D_{\infty}$	(1) Single		
	$\begin{array}{c} T_d:A_2\\ (2) \text{ Double} \end{array}$	C	2	₽ ∞	$D_{\infty}:\Sigma^+$	2	4
	$T_d:2A_1$	<b>n</b>	4		$D_{\infty}$ : $\Sigma^{-}$	c c	4 2
	$T_d: 2A_1$ $T_d: 2A_2$	p c	4		(2) Double	C	2
	$T_d: A_1$ $T_d: A_1 + A_2$	c	4		$D_{\infty}:2\Sigma^+$	c	8
		č	•		$D_{\infty}:2\Sigma^{-}$	c c	4
h	(1) Single				$D_{\infty}:\Sigma^{+}+\Sigma^{-}$	c	4
	$T_h:A_g$	p	2				-
	$T_h: A_u$	c	2	C∞	(1) Single		
	(2) Double				$C_{\infty}:\Sigma$	с	4
	$T_h: 2A_g$	p	4		(2) Double		
	$T_h: 2A_u$	c	4		$C_{\infty}$ : 2 $\Sigma$	с	8
	$T_h:A_g + A_u$	с	4	<i>C</i> ~	(1) Single		
				⊂∞h	$C_{\infty h}: \Sigma_g$	n	2
h	(1) Single $Q_{1} : A$	_	n		$C \approx h \cdot \Sigma g$	p	2
	$O_h:A_{1g}$	p	2		$C_{\infty h}: \Sigma_{u}$ (2) Double	с	2
	$O_h:A_{2g}$	m	1 2			-	4
	$O_h:A_1u$	c m	2		$C_{\infty h}: 2\Sigma_g$	p	4
	$O_h:A_{2u}$ (2) Double	m	1		$C_{\infty h}: 2\Sigma_{u}$	c	4
	$O_h: 2A_{1g}$	-	4		$C_{\infty h}: \Sigma_g + \Sigma_u$	с	4
	$O_h: 2A_{1g}$ $O_h: 2A_{2g}$	p	4 2	K <sub>h</sub>	(1) Single		
	$O_h: 2A_{2g}$ $O_h: 2A_{1u}$	p		<i>"</i>	$K_h:S_g$ Not possible		
	$O_h: 2A_{1u}$ $O_h: 2A_{2u}$	c	4 2		$K_h:S_u$	с	2.1
	$Q_{\mu} \cdot A_{\mu} + A_{\mu}$	p	2		(2) Double	-	-
	$O_h:A_{1g} + A_{2g}$ $O_h:A_{2g} + A_{2g}$	p c	4		$K_h: 2S_u$	с	4
	$O_h: A_{1g} + A_{1u}$ $O_h: A_{2g} + A_{2u}$		4 2	V		-	
	$O_h:A_{1g} + A_{2u}$ $O_h:A_{1g} + A_{2u}$	p	2	K	Yields no derived groups		
	$O_h:A_{2g} + A_{1u}$ $O_h:A_{2g} + A_{2u}$	c c	2				
		1.					

a 10. b 1. c 9. d, e 21: and 22 arbitrary, f 7. 8 5. h 6. i 17. k 18. l 19. m 20. n 16. o 11. P 12. 9 13. r 14. s 15. t Reference 8a, Figure 10. u Reference 8a, Figure 10. v Reference 8a, Figure 10. w meso-Tartaric acid. × dl-Tartaric acid. Y Single asymmetric carbon. These last three examples are trivial since chiral ligands are being treated as two-values points.

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Table III. Summary of Derived Groups

	Single						Double					
i	М	P2	C2	C4	Tot	M	<b>P</b> 2	P4	C2	C4	C8	Tot
$C_1$	0	0	0	1	1	0	0	0	0	0	1	1
$C_{1}$ $C_{S}$ $C_{i}$ $C_{2}$ $C_{3}$ $D_{4}$ $C_{2\nu}$ $C_{3\nu}$ $C_{2h}$ $C_{3h}$	0	1	1	0	2	0	0	1	0	2	0	3 3 3
$C_i$	0	1	1	0	2	0	0	1	0	2	0	3
$C_2$	0	0	1	1	2	0	0	0	0	2	1	3
C <sub>3</sub>	0	0	0	1	1	0	0	0	0	0	1	1
$D_3$	0	0	1	1	2	0	0	0	0	2	1	3
$D_{4}$	0	0	3	1	4	0	0	0	3	6	1	10
$C_{2\nu}$	2	1	1	0	4	0	4	1	3	2	0	10
$C_{3\nu}$	0	1	1	0	2	0	0	1	0	2	0	3
$C_{2h}$	2	1	1	0	4	0	4	1	3	2	0	10
C <sub>3h</sub>	0	1	1	0	2 4	0	0	1	0	2	0	3
$D_{2h}$	2	1	1	0		0	3 12	1	2	1	0	10
$D_{sh}$	6	1	1	0	8	12		1	9	2	0	36
$D_{\gamma d}$	2	1	1	0	4	0	4	1	3	2	0	10
Dad	2	1	1	0	4	0	4	1	3	2	0	10
$S_4$ $T_d$ $T_h$	0	1	1	0	2 2 2	0	0	1	0	2 2 2 2 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	0	3 3 10 3 1
$T_d$	0	1	1	0	2	0	0	1	0	2	0	3
$T_h$	0	1	1	0	2	0	0	1	0	2	0	3
$O_h$	2	1	1	0	4	0	4	1	3	2	0	10
$I_h$	0	1	1	0	2	0	0	1	0		0	3
$I_h$ T O	0	0	0	1	1	0	0	0	0	0	1	1
0	0	0	1	1	2	0	0	0	0	2	1	3 1
I	0	0	0	1	1	0	0	0	0	0	1	
C∞	0	0	1	0	1	0	0	0	0	1	0	1
$D_{\infty h}$ $C_{\infty}$	1	0	1	0	2	0	1	0	1	1	0	3 1
C∞	0	0	0	1	1	0	0	0	0	0	1	1
$D_{\infty}$	0	0	1	1	2	0	0	0	0	2	1	3 3
$C_{\infty h}$	0	1	1	0	2	0	0	1	0	2	0	3
Kh	0	0	1	0	1	0	0	0	0	1	0	1
Κ	0	0	0	0	0	0	0	0	0	0	0	0
Total:	19	16	26	10	71	12	37	16	31	49	10	155

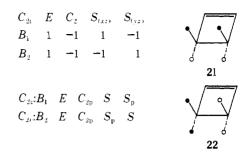
metry point groups is used. For example,  $C_{2\nu}$  is representative of all  $C_{n\nu}$ , *n* even. The information given for each group is its name, type (m = meso, p = pseudochiral, c = chiral), number of isomers, and the location or name of any examples. There is a simple relationship:

## $ne = 2^{s+1}$

where *n* is the number of isomers, *e* is the number of types of symmetry elements the structure has, and *s* is the number of sets of chiral ligands. Some of the infinite groups yield no derived groups, or groups that cannot be realized in any chemical structure and are included only for completeness (see Table III for summary of derived groups). Since chiral ligands are being considered as two-valued points, the group  $K_h S_u$  represents a single asymmetric carbon atom.

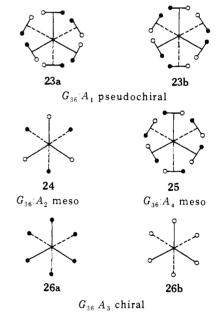
A number of tables of single and double antisymmetry groups exist  $S_w$  (also called magnetic symmetry groups).<sup>11</sup> The present derivation is necessary because of some important differences between the two problems, which change the number and kind of groups obtained. It has already been noted that the operations are labeled differently. Furthermore, double antisymmetry groups of the type  $C_{2h}:A_{\mu}$ +  $B_u$  and  $C_{2h}:B_u + A_u$  are considered to be different. In the present case, such a distinction leads to constitutional isomerism. Antisymmetry groups resulting from the totally symmetric representation are trivial. In the present case, they yield pseudochiral structure, the major form of isomerism here. Finally, degeneracies which occur in the antisymmetry groups are removed in the present case. For example, the groups  $C_{2\nu}:B_1$  and  $C_{2\nu}:B_2$  (Scheme V) are considered as a degenerate pair since the planes of symmetry differ only by labels. However, these two groups can be represented by different chemically realizable stereoisomers (21 and 22) and therefore must be considered different. The assignment of each example to its pseudochirality group is arbitrary, unless the molecular orientation is fixed.

Scheme V



Groups with three or more sets of chiral ligands would be derived similarly. There would be about 400 such groups with three sets of chiral ligands. Such chemical structures are easily realizable, particularly as cyclic peptides. It is interesting that there can be no meso structure with three or more sets of chiral ligands. This can be easily proved by noting that meso structures occur only for  $D_{nh}$  groups with two sets of chiral ligands. A meso structure can result only if the homomorphism of the symmetry group onto the chirality change group is onto with respect to both rotation and reflection symmetry. Since  $D_n$  has only two generators, it cannot be mapped (epimorphically) onto  $C_2 \times C_2 \times C_2$ . Hence no meso structure is possible.

A further generalization can be made by considering nonrigid structures.<sup>13,14</sup> Longuet-Higgins<sup>15</sup> has given the nonrigid symmetry group for ethane, symbolized here as  $G_{36}$ . The possible pseudochirality groups are determined in the same manner as above, by imposition onto the one-dimensional real irreducible representations of the group. There are four of these, and examples are provided by structures 23-26. Longuet-Higgins has given the character table and the labels for the irreducible representations.<sup>15</sup>



## **Overview and Extensions**

An overview of this work is hopefully ascertainable from the following discussion. Consider the three configurations of chiral ligands, A, B, and C. In configuration C, the solid

0 <b>0</b>	0 O	• •
0 0	• •	• •
А	В	С

vertical line is above the plane, and the dotted horizontal line is below the plane. The symmetry of these structures can be specified in several ways. The first and simplest is to specify just the rotation symmetry. A has D4 rotation symmetry, **B** has  $C_2$  rotation symmetry, and **C** has  $D_2$  rotation symmetry. Knowing only rotation symmetry is sufficient to describe some properties of chemical structures. For example, the number of isomers is determined using just the rotation symmetry.<sup>16</sup> The  $\pi$ -molecular orbitals of benzene can be determined from just the rotation group  $C_{6}$ .<sup>17</sup> Specifications of only rotation symmetry is, however, insufficient to determine other properties such as chirality. This leads to the next higher level of symmetry specification, namely point group symmetry. A has  $D_4$  symmetry, B has  $C_{2h}$  symmetry, and C has  $D_{2d}$  symmetry. From these specifications it is apparent that A exists in two enantiomeric forms, and B and C are superimposible on their mirror images. The point group description is a great deal more useful than the rotation group description, and most applications of group theory to chemistry require it. The point group designation is, however, inadequate to ascertain pseudochirality.

The present work extends this sequence by one. Of the examples given, A is  $D_{4h}:A_{1u}$ , B is  $D_{2h}:B_{2g}$ , and C is  $D_{2d}:$  $A_1$ . This indicates that B is meso and C is pseudochiral and exists in two pseudoenantiomeric forms. The information gained (pseudochirality) by specifying the pseudochiral symmetry and point group symmetry instead of just the point group symmetry is analogous to the information gained (chirality) by specifying point group symmetry instead of just rotation symmetry.

This sequence can be extended and the work generalized in any number of ways. What is needed is a designation of the group C in (1). For example, the members of this group could include operations which change ligand constitution. The structure 27 yields a representation for  $C_{6h}:B_g + E_{1g}$ .



The development here is somewhat similar to the treatment of polychromatic symmetry.11,18

The group C could also include operations which reverse the chirality of ligands *individually*, rather than all at once. If one also includes all permutations of constitutionally identical ligands (not just those which represent the point group), one obtains the hyperoctahedral groups used by Mead<sup>19</sup> to obtain chirality functions for molecules with chiral ligands. Pseudochirality groups are subgroups of these hyperoctahedral groups. One could classify stereoisomers more generally by finding other subgroups of the hyperoctahedral group which leave various stereoisomers invariant. For example, structure 28 is left invariant by the



permutation  $(\overline{1})(\overline{2})(34)$  which is the operation of reflection in the vertical plane followed by reversal of the chirality ligand s1 and 2. This structure (28) is also invariant to the  $C_3$ rotation (123)(4). All products of these two permutations,  $(123)(4)(\overline{1})(\overline{2})(34) = (\overline{12}34)$ , etc., generate a group of order 24 which is isomorphic to  $T_d$  and is a subgroup of the hyperoctahedral group. Other generalizations are possible by modifying the combination of the groups. The diagram shown in (1) is for the direct product, which is a rather trivial extension of C by  $S^{20a}$  More generalized extensions are represented by the diagrams 2 and 3. Some nonrigid symmetry groups can be expressed as semidirect products.<sup>20b</sup>

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$$C \longrightarrow Ux, S \rightleftarrows S$$
 (2)  
general extension

$$C \longrightarrow CextS \longrightarrow S \tag{3}$$

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### **References and Notes**

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- (12)One of the referees has objected to the use of the word pseudochirality in preference to pseudoasymmetry. It is not the purpose of this paper to suggest standard chemical nomenclature, nevertheless, the term pseudochirality is used deliberately for the mathematical and semantic reasons given in the text. Structures for which the distinction between the descriptions pseudochiral and pseudoasymmetric is meaningful (namely those with nontrivial rotation symmetry) will be relatively rare but have been suggested in the existing literature (see footnotes to Table II).
- (13) J. E. Leonard, Ph.D. Thesis, submitted May 24, 1971, California Institute of Technology. This reference includes an interesting discussion of nonrigid symmetry groups and meso structures. Of particular interest is the derivation of a group for meso-biphenyls which have no achiral conformations easily accessible.14
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  (21) Explanation of symbols and headings is as follows: the supergroup is the
- point group from which the pseudochirality group is derived; single pseudochirality groups are those for one set of chiral ligands; and double pseudochirality groups are for two sets of chiral ligands. This corre-sponds to earlier terminology for antisymmetry groups and has nothing to do with double-point groups. The derived group designation G: R encompasses the supergroup G, and the representation(s) R imposed as discussed in the text. The *type* refers to the isomer type for which c is chiral, p Is pseudochiral, and m Is meso. Superscripted letters on the number of isomers refer to footnotes giving the location of examples for some of the groups
- (22) H. Hirschmann and Kenneth R. Hanson, J. Org. Chem., 36, 3293 (1971). Their structures 3a and 3b are C<sub>2V</sub>:B<sub>1</sub> + B<sub>2</sub> chiral; 22 is C<sub>2h</sub>:B<sub>g</sub> meso; 28 is D2:B1 chiral.
- (23) This group is not physically realizable because there cannot be on planes in a structure containing discrete chiral units. This is another difference from dichromatic antisymmetry in which infinite symmetry planes are possible